

# Unit II

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- Multiplication of dislocations
- **Techniques to observe dislocation**
- Dislocation - point defects interactions
- Intersection of Dislocations
- Dislocations pile up.
- Deformation of single and polycrystalline materials
- Grain boundaries.
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  - Surface tension of the grain boundary
  - Strengthening from grain boundaries
  - Hall-petch equation
  - Yield point phenomenon.

# Dislocation in FCC lattice

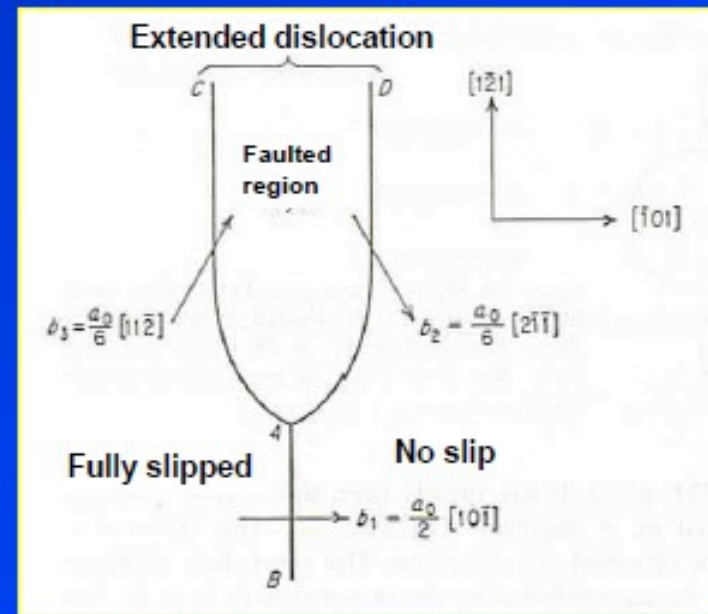
- Slip occurs in the **FCC** lattice on the **{111}** plane in the **<110>** direction and with a Burgers vector  **$(a/2)[110]$** .
- The **{111}** planes are stacked on a close packed sequence **ABCABC** and vector  **$b = (a_0/2)[101]$**  defines one of the observed slip direction, which can favourably energetically **decompose into two partial dislocations**.

$$b_1 \rightarrow b_2 + b_3$$

$$\frac{a_0}{2} [101] \rightarrow \frac{a_0}{6} [2\bar{1}\bar{1}] + \frac{a_0}{6} [11\bar{2}]$$

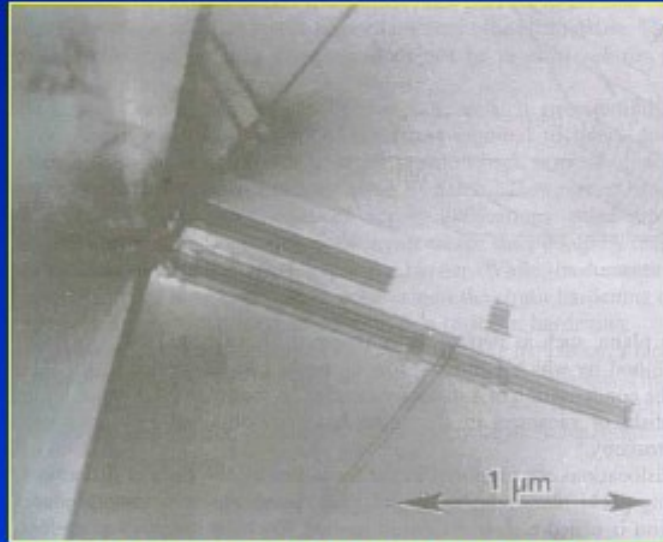
**Shockley partials**

This **Shockley partials** creates a stacking fault **ABCAC/ABC**.

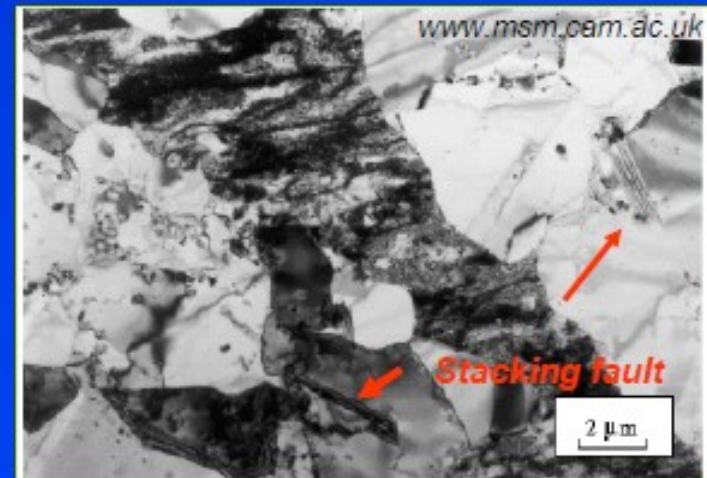
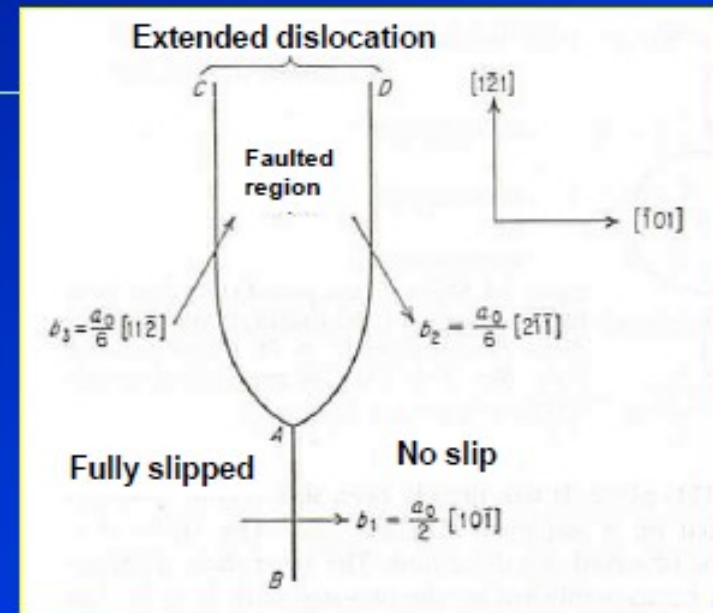


## Dissociation of a dislocation into two partial dislocations

- The combination of the two partials **AC** and **AD** is known as an **extended dislocation**.
- The region between them is a **stacking fault** which has undergone slip.
- The equilibrium of these partial dislocations depends on the stacking fault energy.



Group of stacking fault in 302 stainless steel stopped at boundary



# Stacking faults

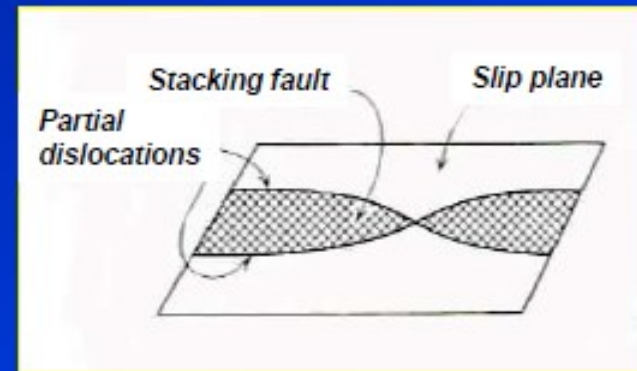
*The wider region between partial dislocation, the lower stacking fault energy*

• Characteristics of metals with low SPF:

- 1) Easy to strain harden
- 2) Easy for twin annealing to occur
- 3) Temperature dependent flow stress

• **Aluminium** – high stacking fault energy  
→ more likely to cross slip.

• **Copper** – lower stacking fault energy → cross slip is not prevalent.



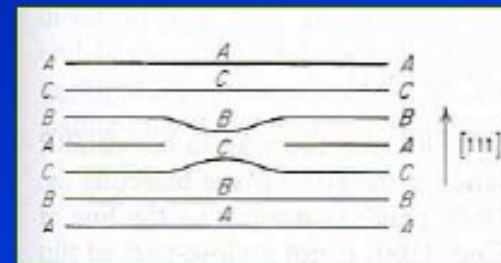
*Model of a stacking fault.*

**Typical values of stacking fault energy**

Metal	Stacking fault energy (mJ m <sup>-2</sup> )
Brass	<10
303 stainless steel	8
304 stainless steel	20
310 stainless steel	45
Silver	~25
Gold	~50
Copper	~80
Nickel	~150
Aluminium	~200

## Frank partial dislocations

**Frank partial dislocations** are another type of partial dislocation in **FCC** lattice, which provide **obstacles** to the movement of other dislocations.



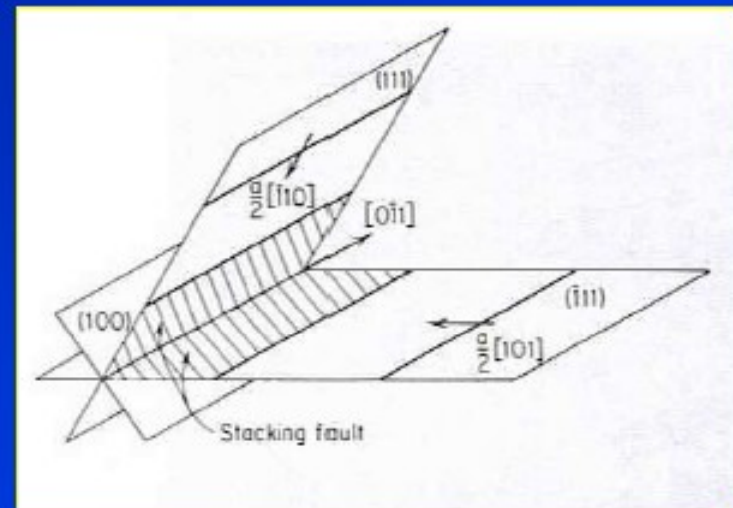
**Frank partial dislocation or sessile dislocation.**

- A set of **(111)** plane (viewed from the edge) has a missing middle **A** plane with a **Burgers vector**  $(a_0/3) [111]$  perpendicular to the central stacking fault.
- Unlike perfect dislocation, **Frank partial dislocation** cannot move by glide (**sessile dislocation**) but by diffusion of atom.

## Lomer-Cottrell barrier

Intersection of  $\{111\}$  plane during duplex slip by glide of dislocations is called **Lomer-Cottrell barrier**.

**Ex:** consider two perfect dislocations lying in different  $\{111\}$  planes and both parallel to the line of intersection of the  $\{111\}$  plane.



**Lomer-Cottrell barrier**

$$\frac{a_o}{2} [101] + \frac{a_o}{2} [\bar{1}10] \rightarrow \frac{a_o}{2} [011]$$

*The new dislocation obtained has reduced energy.*

# Dislocations in HCP lattice

- Slip occurs in the **HCP** lattice on the basal **(0001)** plane in the  **$\langle 11\bar{2}0 \rangle$**  direction.
- The basal **(0001)** plane is close packed of a sequence **ABABAB** and a Burgers vector  **$\mathbf{b} = (a_o/3)[11\bar{2}0]$** .
- Dislocations in the basal plane can reduce their energy by dissociating into Shockley partials according to the reaction.

$$\frac{a_o}{3}[11\bar{2}0] \rightarrow \frac{a_o}{3}[10\bar{1}0] + \frac{a_o}{3}[01\bar{1}0]$$

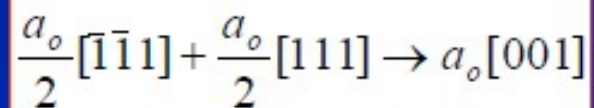
The **stacking fault** produced by this reaction lies in the basal plane, and the extended dislocation which forms it is confined to glide in this plane.



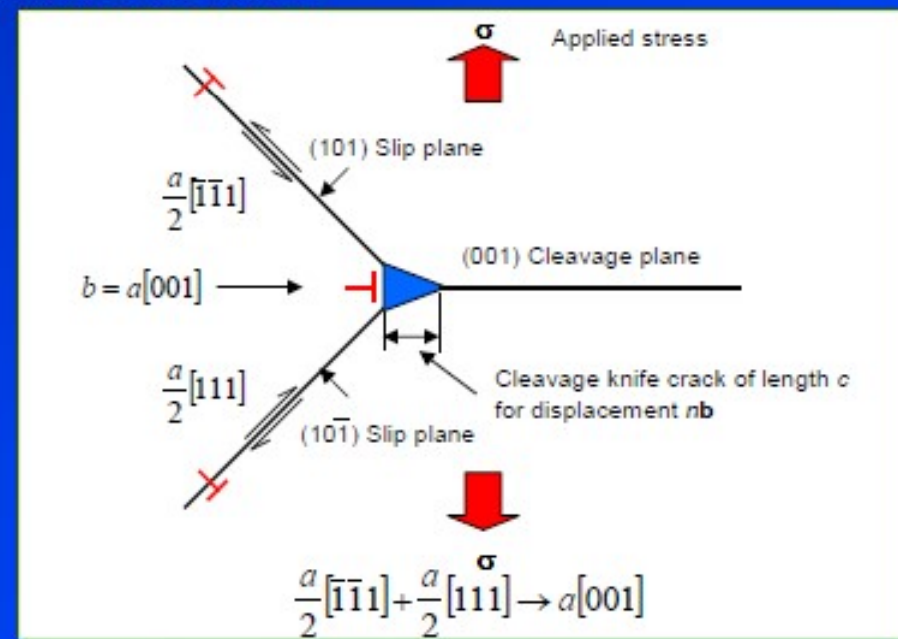
# Dislocations in BCC cubic lattice

- Slip occurs in the **BCC** lattice on  $\{110\}$ ,  $\{112\}$ ,  $\{123\}$  planes in the  $\langle 111 \rangle$  direction and a **Burgers vector**  $\mathbf{b} = (a_o/2)[111]$ .

**Cottrell** has suggested a dislocation reaction which appears to cause **immobile dislocations**. ( $a_o/2[001]$  in iron)  $\rightarrow$  leading to a crack nucleus formation mechanism for brittle fracture.



the dislocation is **immobile** since the **(001)** is not a close-packed slip plane, the **(001)** plane is therefore the **cleavage plane** when brittle fracture occurs.



# Sources of Dislocation

- Except single crystal like whiskers, all other crystals have dislocations.
- The dislocations are produced during the growth of the crystal from the melt or vapour phase. Temperature and composition gradients may produce misalignments and produce dislocation in grain boundaries.
- Dislocations can be generated by thermal and mechanical stress near the particles, twins, cracks or surface flaws.
- Annihilation of vacancies produces dislocation like Frank partial dislocations in FCC metals.
- Plastic deformation produces or multiplying the dislocations. Cold worked metals have many dislocations.

# Dislocation Multiplication

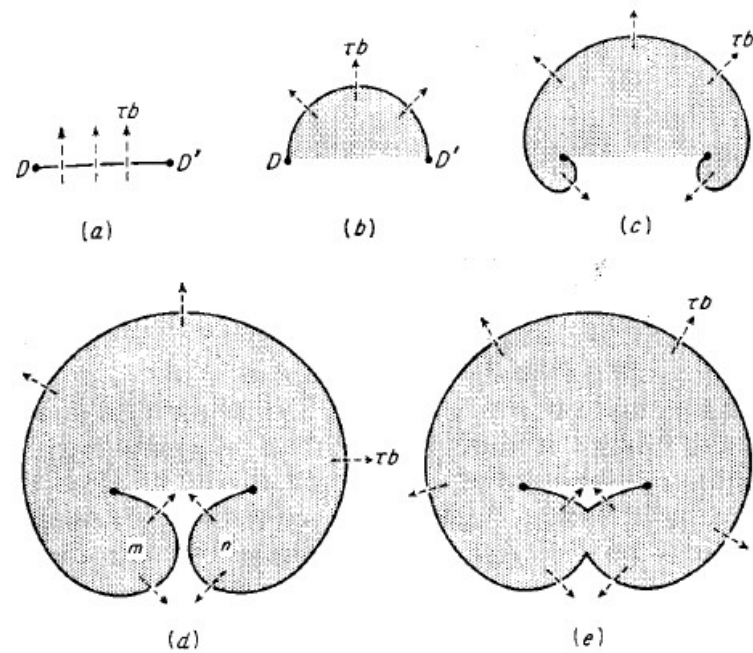
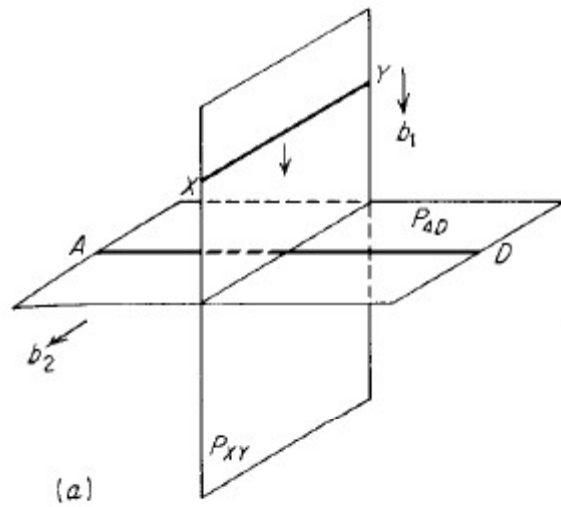


Figure : Frank-Read source for dislocation multiplication.

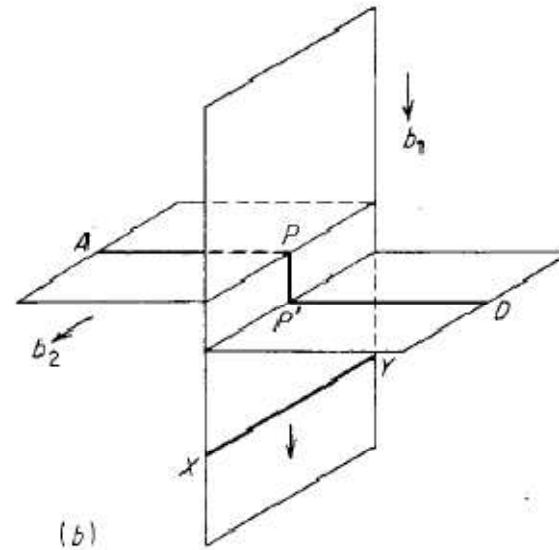
# Intersection of Dislocation

- The intersection of two dislocations produces a sharp break, a few atom spacing in length, in the dislocation line. These breaks are two types.
  - **Jog** : A sharp break in the dislocation moving it out of the slip plane.
  - **Kink**: A sharp break in the dislocation line which remains in the slip plane.
- Possibilities
  - Intersection of two edge dislocations with perpendicular Burgers vectors
  - Intersection of two edge dislocations with parallel Burgers vectors
  - Intersection of an edge dislocation and screw dislocation
  - Intersection of two screw dislocation

# Intersection of two edge dislocations with perpendicular Burgers vectors



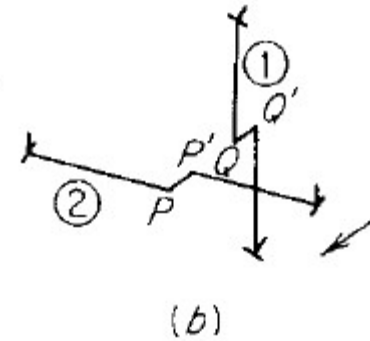
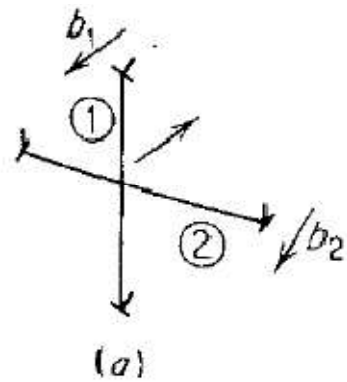
Before intersection



After intersection

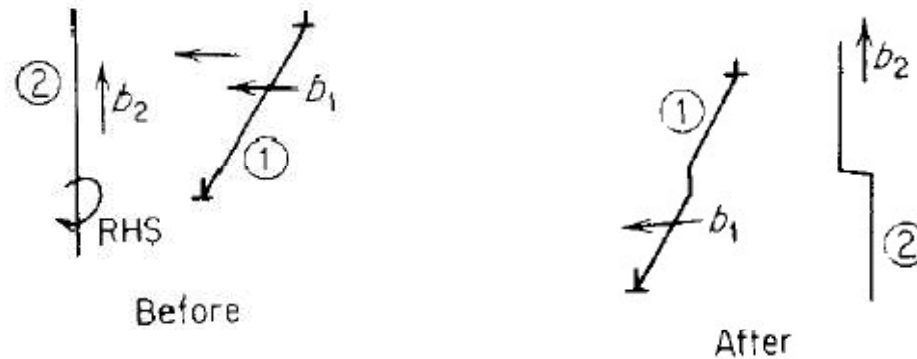
Produce jog  $PP'$  has edge nature

# Intersection of two edge dislocations with parallel Burgers vectors



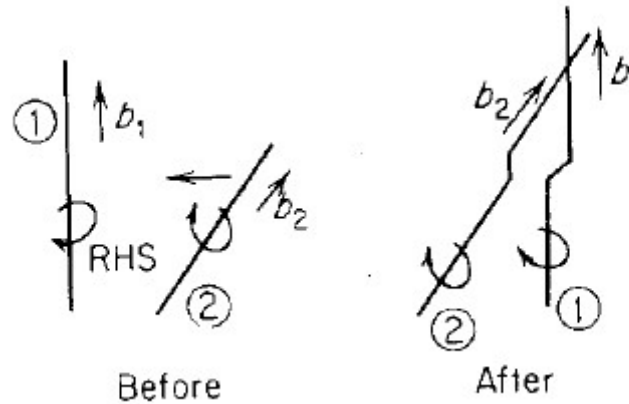
Both kinks are in screw nature.

# Intersection of an edge dislocation and screw dislocation



Produce jog in edge dislocation with edge nature and a kink in screw dislocation with edge nature.

# Intersection of two screw dislocation

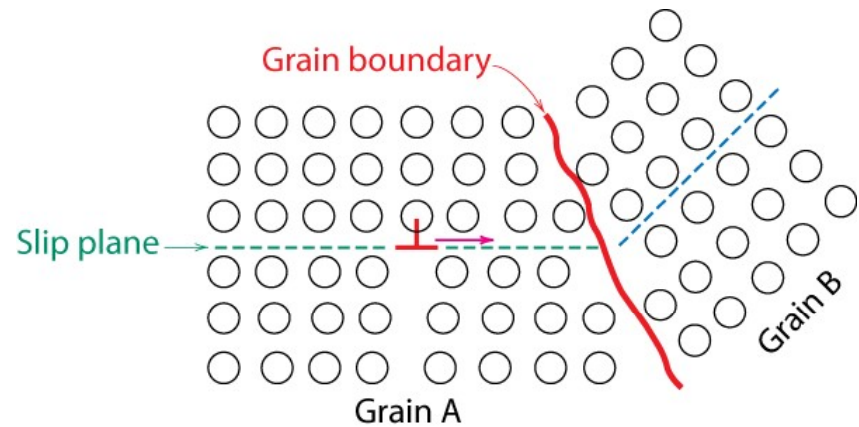


Intersection of two screw dislocations develops a jog in edge nature.



# Grain size Strengthening

- Grain boundary act as barrier for dislocation motion: slip plane discontinues or change orientation.
- Dislocations gliding on a slip plane are unable to cross the boundary but get piled up against it. This happens more often in a fine grained material.
- **Small angle grain boundaries** are not very effective in blocking dislocations.



- **High-angle grain boundaries** block slip and increase strength of the material.

The yield strength of a poly crystalline material is a function of its grain size as given by the **Hall petch equation**

$$\sigma_y = \sigma_i + kd^{-1/2}$$

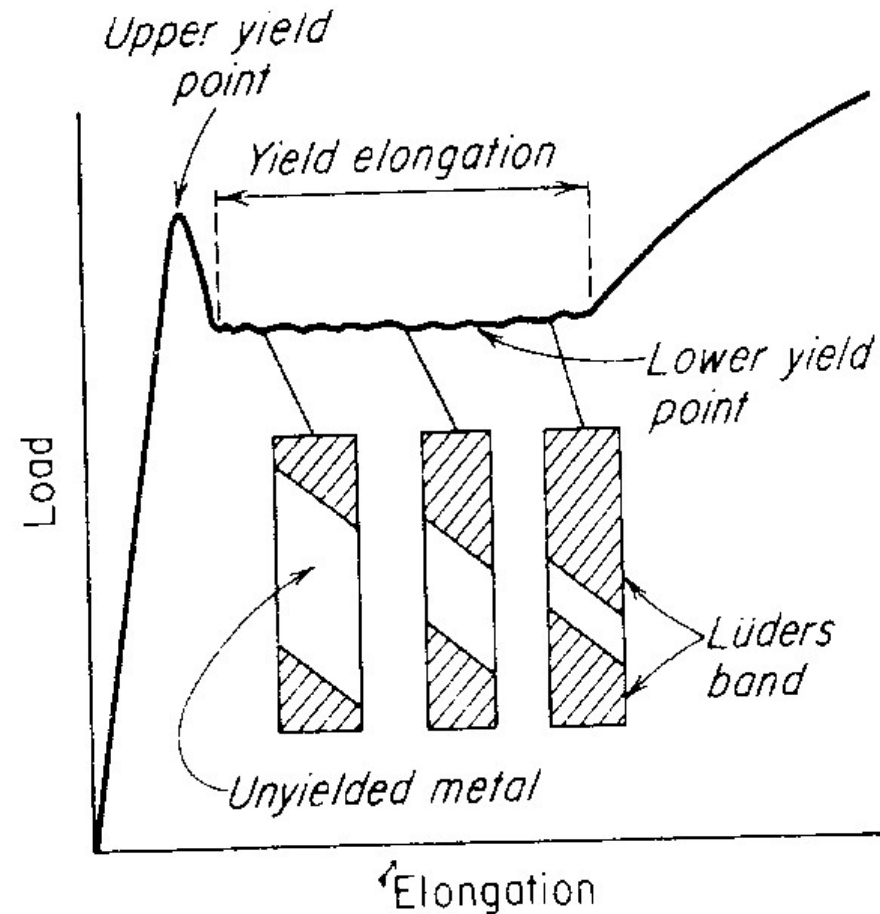
$\sigma_i$  the yield strength of the material where there are no grain boundaries (single crystal),

$k$  is the relative hardening contribution of the grain boundaries or the Hall Petch constant

$d$  is the grain diameter, which can be derived from the ASTM grain size number.

# Yield Point Phenomenon

- The load at which sudden drop occurs – Upper Yield Point
- The average constant load – Lower yield point
- The elongation which occurs at constant load – yield point elongation
- Discrete band formation in the material during yield – Lüders band



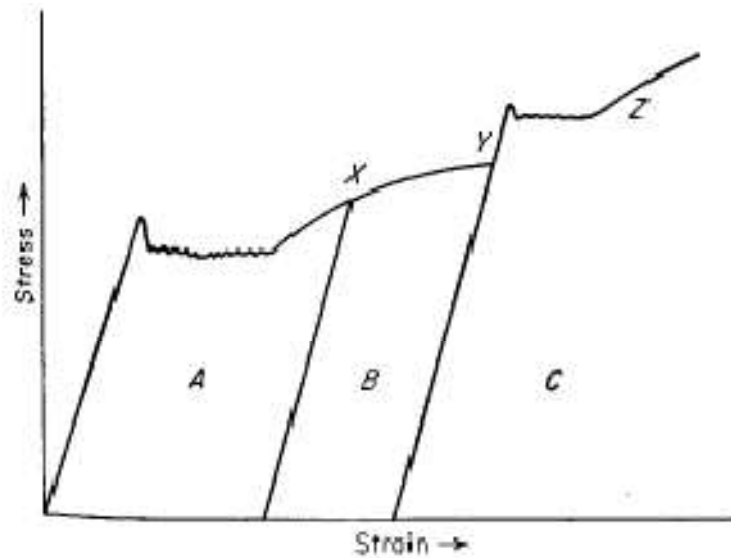
**Due to interstitial or substitutional impurities**

## Yield point phenomena observed in

- Steel
- Mo, Ti and Al alloys
- Single crystals of Fe, Cd, Zn, brass, Al

# Strain Ageing

- A metal hardens as a result of ageing after plastic deformation is called strain ageing.

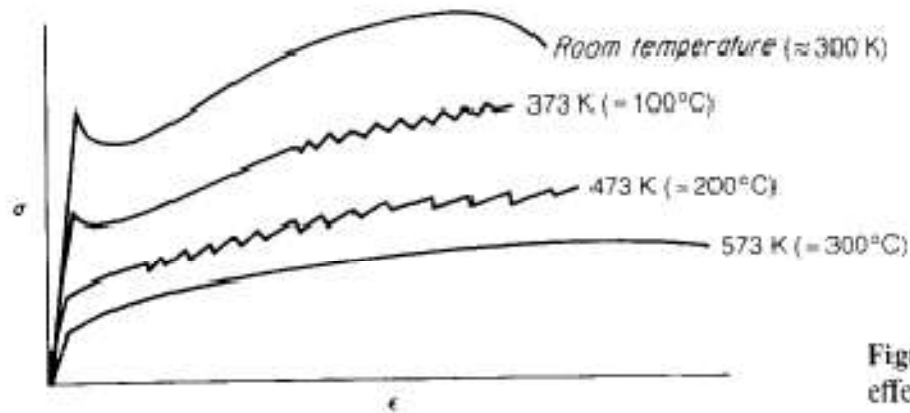


To control the strain ageing,

- Lower the amount of carbon and nitrogen.
- Alloying element Al, V, Ti, and boron are added to convert carbon and nitrogen as stable carbides or nitrides.
- Deform the metals immediately before it can age.

# Dynamic strain-ageing

- Strain ageing is associated with the occurrence of serrations in the stress-strain curve
- Dynamic strain-ageing behaviour or Portevin-LeChatelier effect



For plain C steel

- The solute atoms are diffuse in the specimen at a rate faster than the speed of the dislocations and arrest the dislocations
- Due to mechanical twinning during deformation or stress assisted martensitic transformations

- PCS discontinuous yielding / dynamic strain-ageing occurs in the temperature region of 500 to 650 K. This temperature region is known as ***blue brittle region***